APPROVED FOR RELEASE: Thursday, May 23, 2002 CIA-RDP91-00772R000100350050-4

AEC-tr-4482(p.1488-1529) Uncl.

DIRECT ABSORPTION OF RADIOACTIVE ISOTOPES IN WATER BY SALT WATER AND FRESH WATER FISH.
Tetsuo Tomiyama and Kunio Kobayashi.

C-23 P NSA

N-4

APPROVED FOR RELEASE: Thursday, May 23, 2002 CIA-RDP91-00772R000100350050-4

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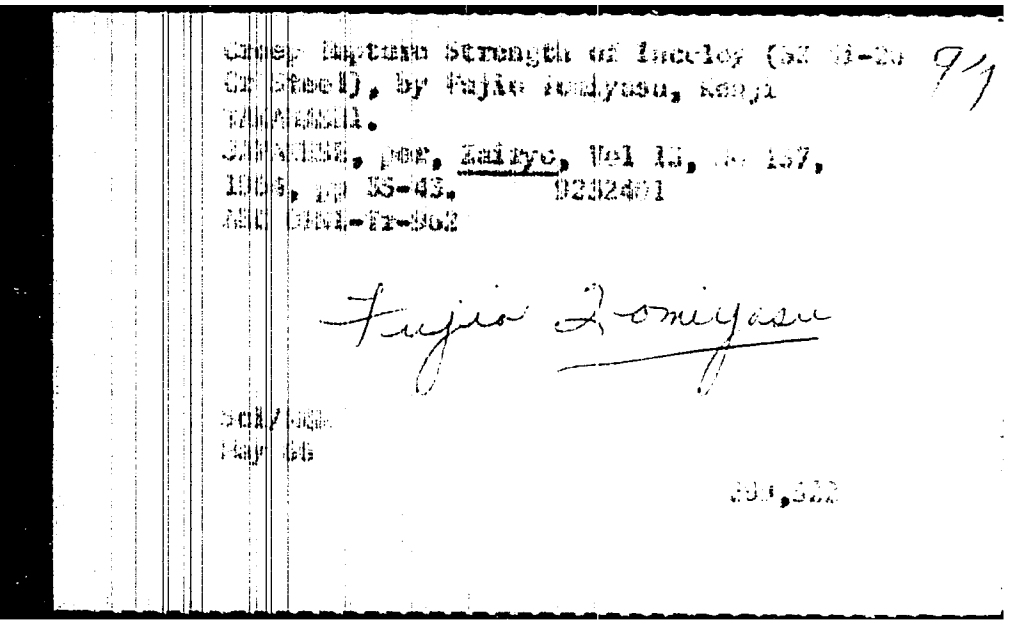

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Fujio Tomiyasu

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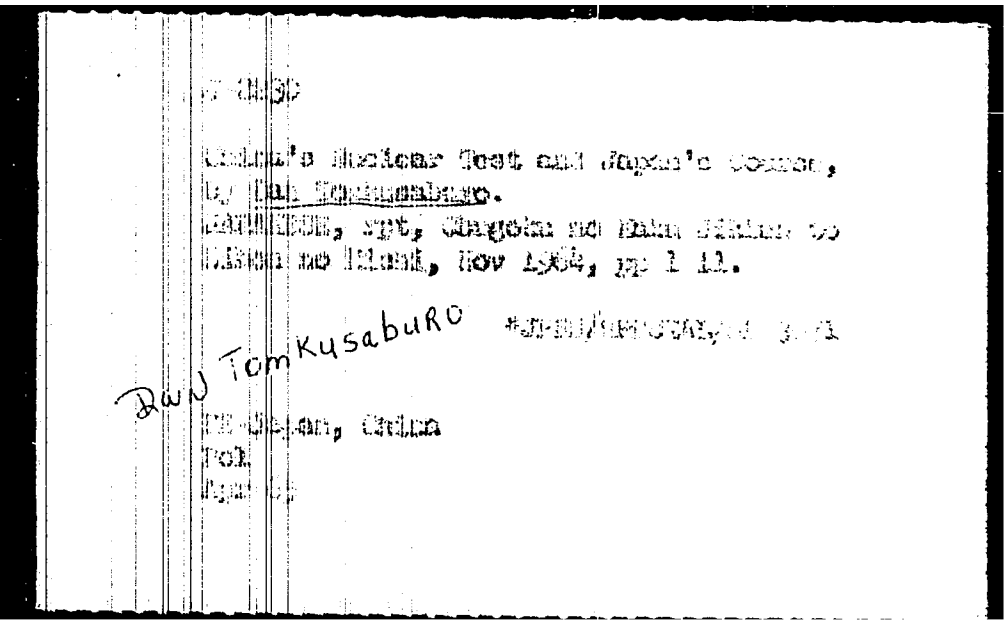

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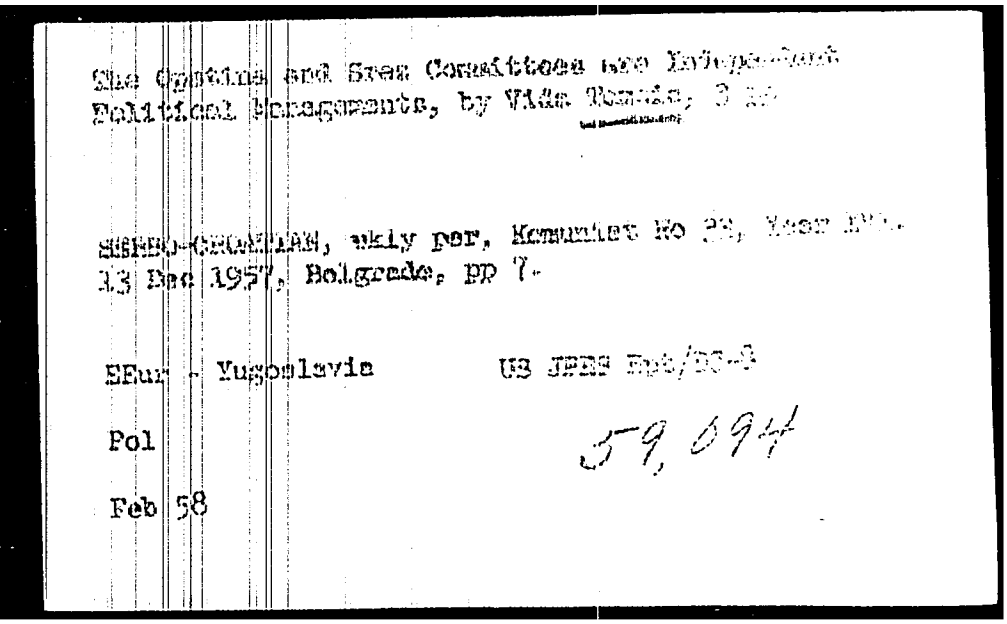

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SLA 59-10300

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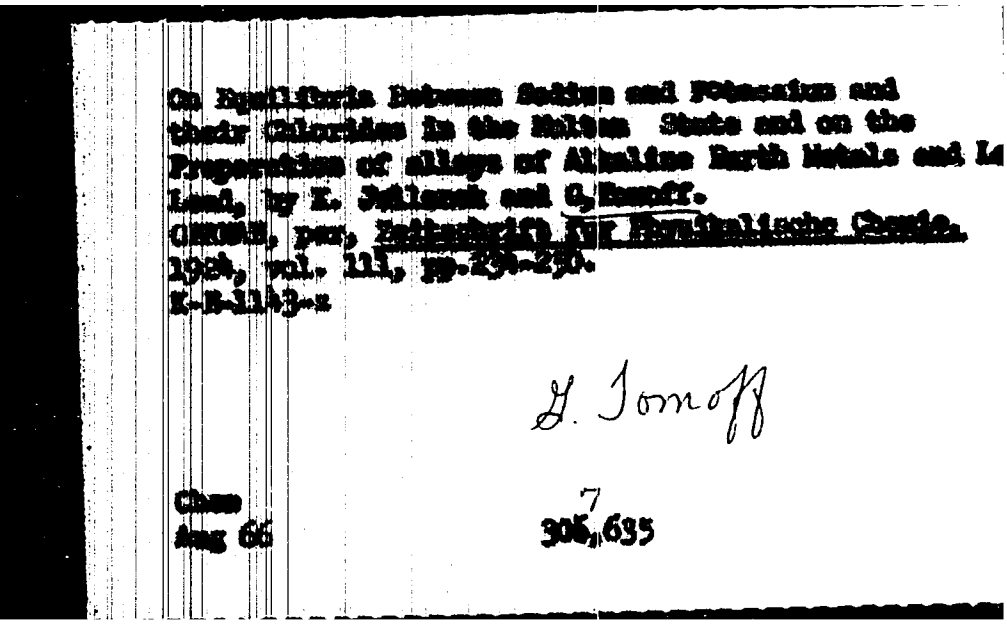

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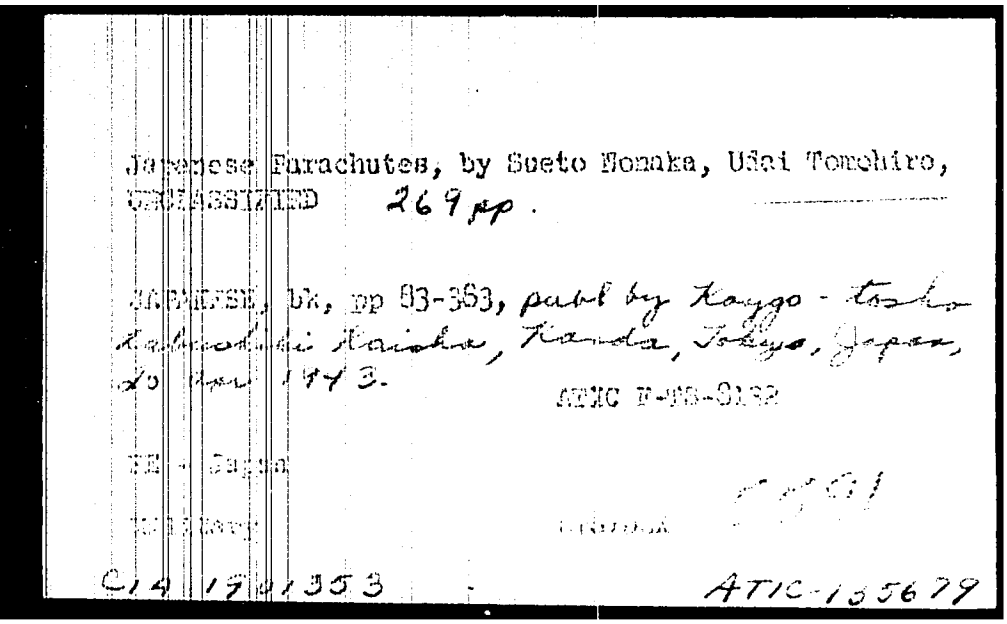

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QUANTUM MECHANICS, tr. by Koshiba. 1962, 326p
Order from Interscience Publishers, Inc., 250 Fifth Ave., New York, 1, N. Y. $12.50

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DESCRIPTORS: *Quantum mechanics

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63-18809

Tomonari, Akihisa.
AN IMPROVEMENT ON THE METHOD FOR SPECTROPHOTOMETRIC DETERMINATION OF TRACES OF CHLORIDE ION. [1963] 7p. (figs. tables omitted) 6 refs.
Order from SLA $1.10 63-18809

Trans. of Nihon Kagaku Zasshi (Japan) 1962, v. 83, no. 6, p. 693-695.

DESCRIPTORS: *Chlorides, Ions, Chemical analysis, *Spectroscopy, *Colorimetric analysis, Calibration, *Iron compounds, *Perchlorates, *Mercury compounds, *Thiocyanates, Nitrates.

An iron perchlorate solution and a mercuric thiocyanate solution were added to a specimen solution and the chloride ion was determined by measuring the absorbency of an orange-colored aqueous solution at
(Chemistry--Analytical, TT, v. 10, no. 12) (over)

I. Tomonari, A.

Office of Technical Services

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APPROVED FOR RELEASE: Thursday, May 23, 2002 CIA-RDP91-00772R000100350050-4

460 m(mu). It was possible to determine from 0.1 to 15 ppm. of chloride ion speedily and accurately. (Author)

PB- 63-18809

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Tomonari, Akihisa.
SPECTROPHOTOMETRIC DETERMINATION OF TRACES OF CHLORIDE IN LARGE AMOUNTS OF BROMIDE. [1963] 10p. (figs. tables omitted) 7 refs.
Order from SLA $1.10 63-18715

Trans. of Nihon Kagaku Zasshi (Japan) [1962] v. 83, no. 6, p. 696-698.

DESCRIPTORS: *Chlorides, *Bromides, Oxidation, Spectrophotometers, Quantitative analysis.

A method for the determination of traces of chloride in large amounts of bromide was studied. An aqueous solution containing a fixed quantity of bromide ion was oxidized to potassium permanganate at the time when it was heated in nitric acidity. Conditions for the quantitative oxidation of bromide ion to bromine were sought. After removing bromine by heating the chloride (Chemistry--Analytical, TT, v. 10, no. 11) (over)

63-18715

1. Tomonari, A.

Office of Technical Services

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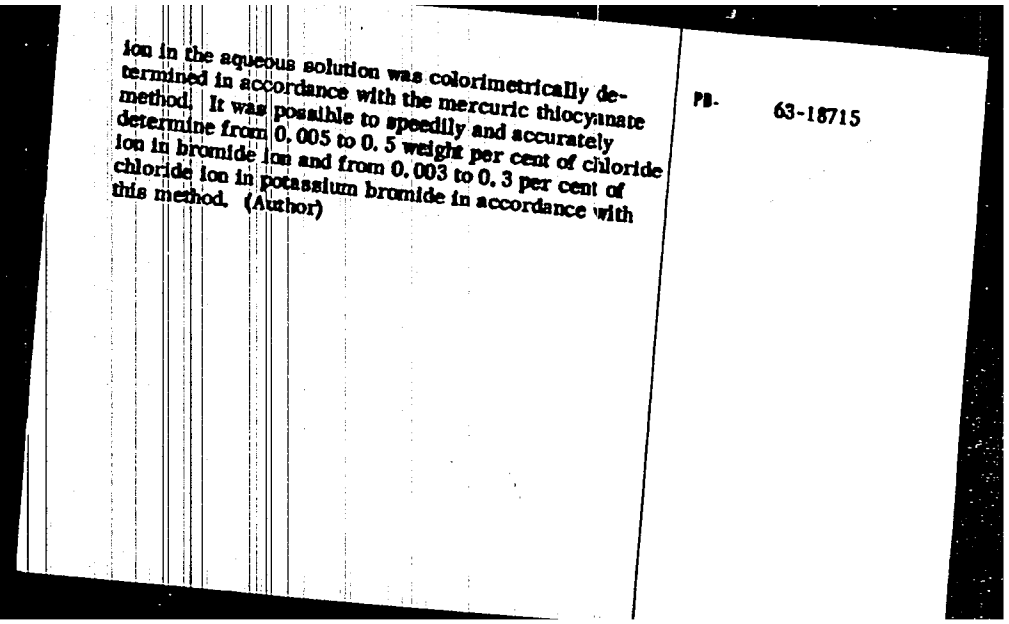

ion in the aqueous solution was colorimetrically determined in accordance with the mercuric thiocyanate method. It was possible to speedily and accurately determine from 0.005 to 0.5 weight per cent of chloride ion in bromide ion and from 0.003 to 0.3 per cent of chloride ion in potassium bromide in accordance with this method. (Author)

PB- 63-18715

APPROVED FOR RELEASE: Thursday, May 23, 2002 CIA-RDP91-00772R000100350050-4

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On Aromatic Antimony Trichloride Complex Report No. 4., by H. Tozono, 17 pp.

JAPANESE, per, Kagaku Kenkyujo Hokoku (Reports of the Institute for Chemical Research), Vol XXIV, March 1951, pp 54-61.

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On Aromatic Antimony Trichloride Complex. Report I., by Hajime Tomono, 15 pp.

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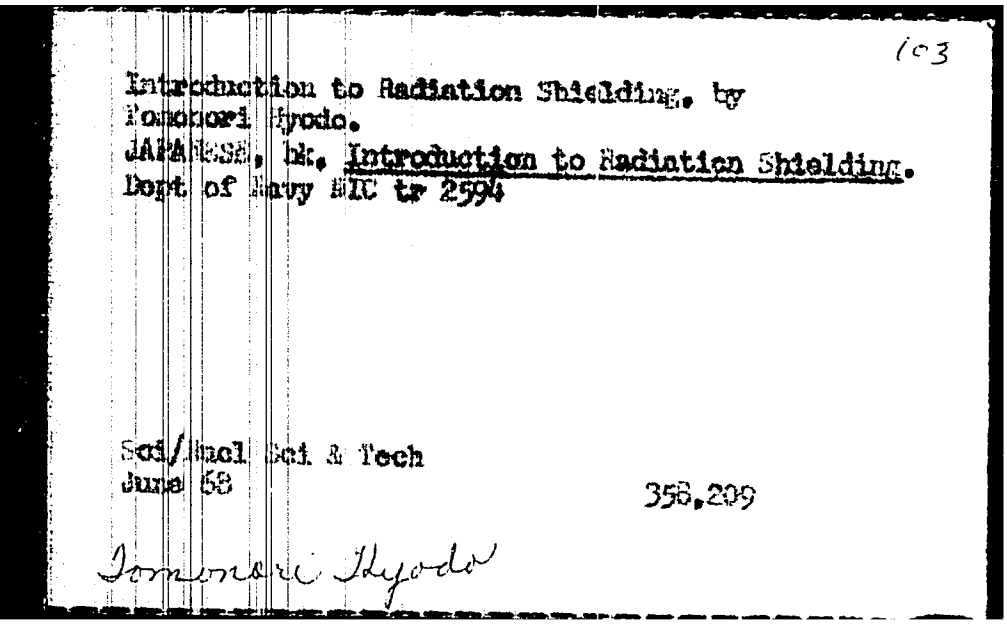

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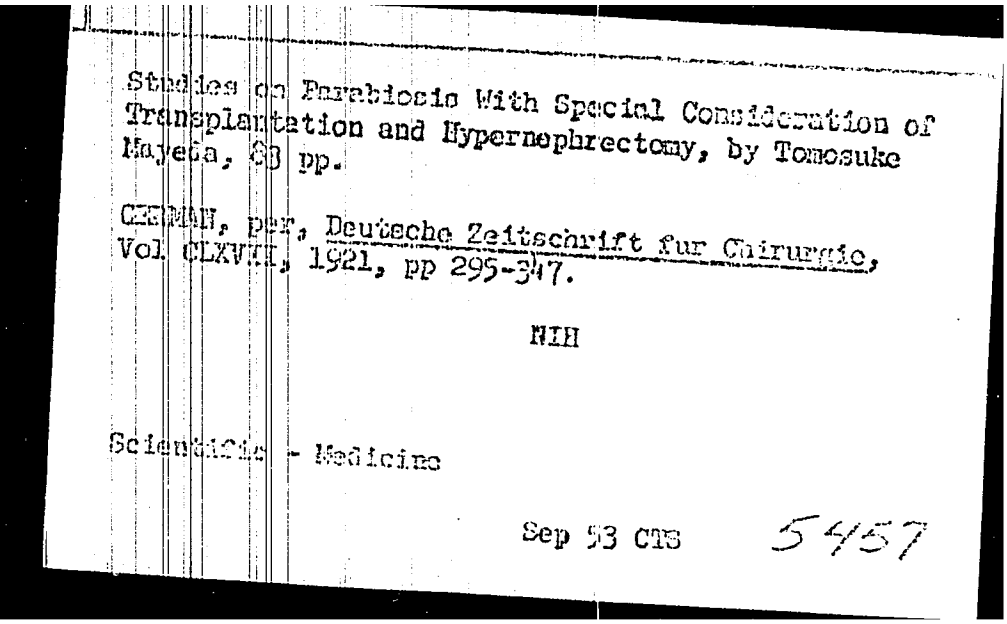

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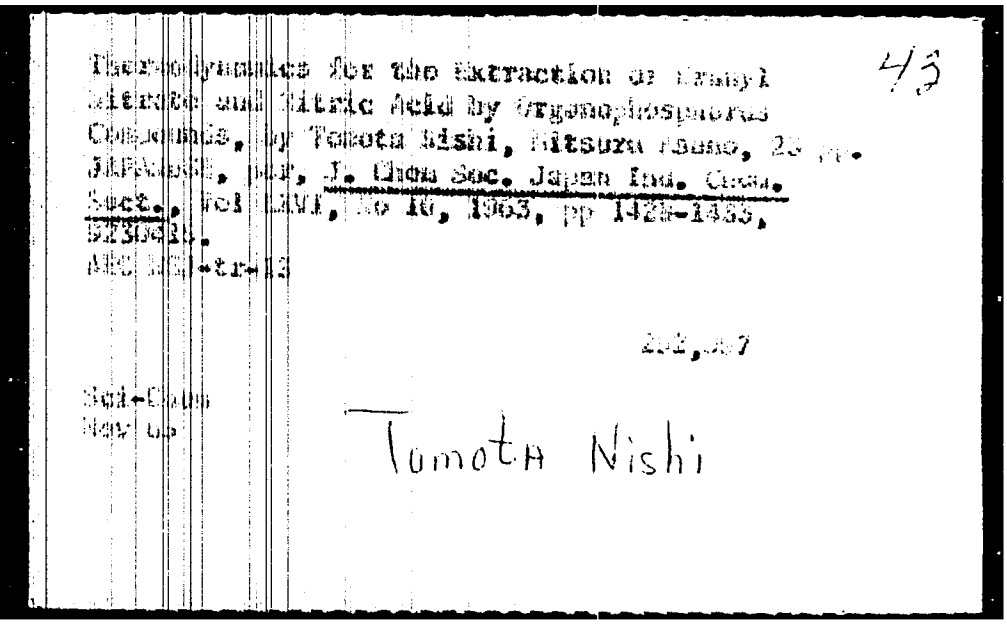

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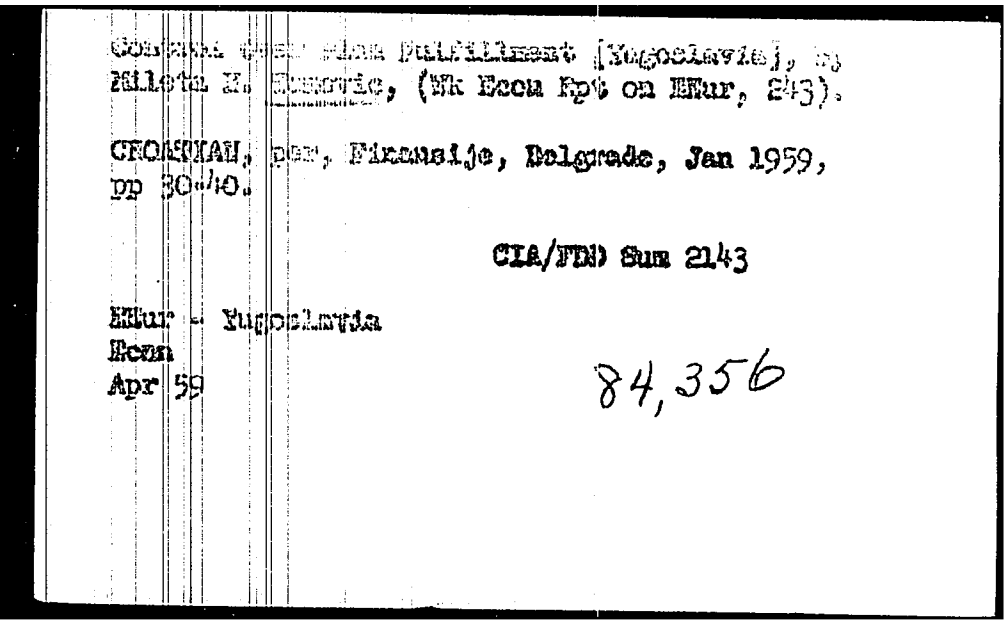

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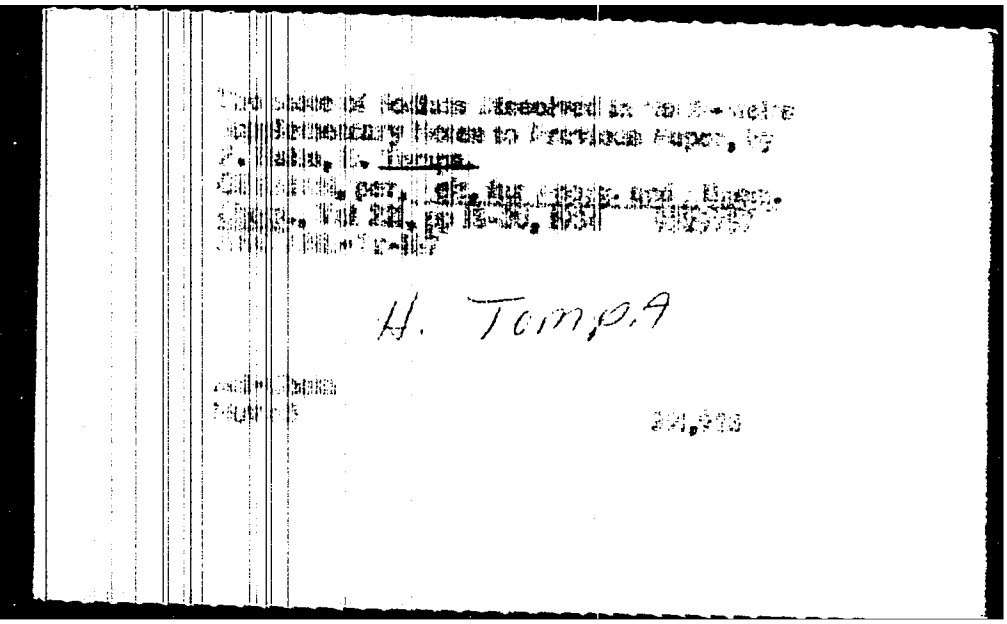

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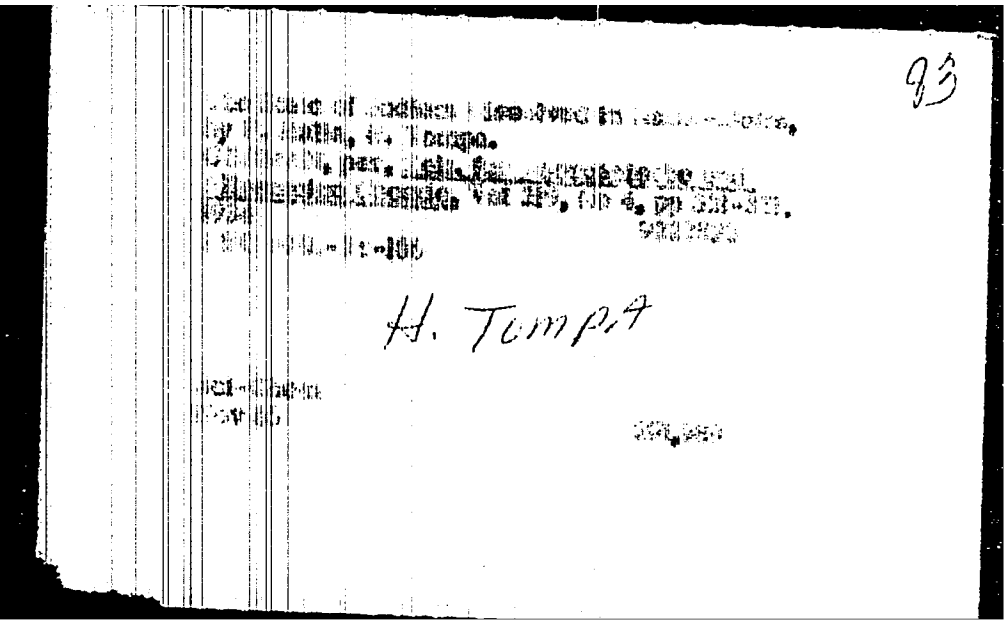

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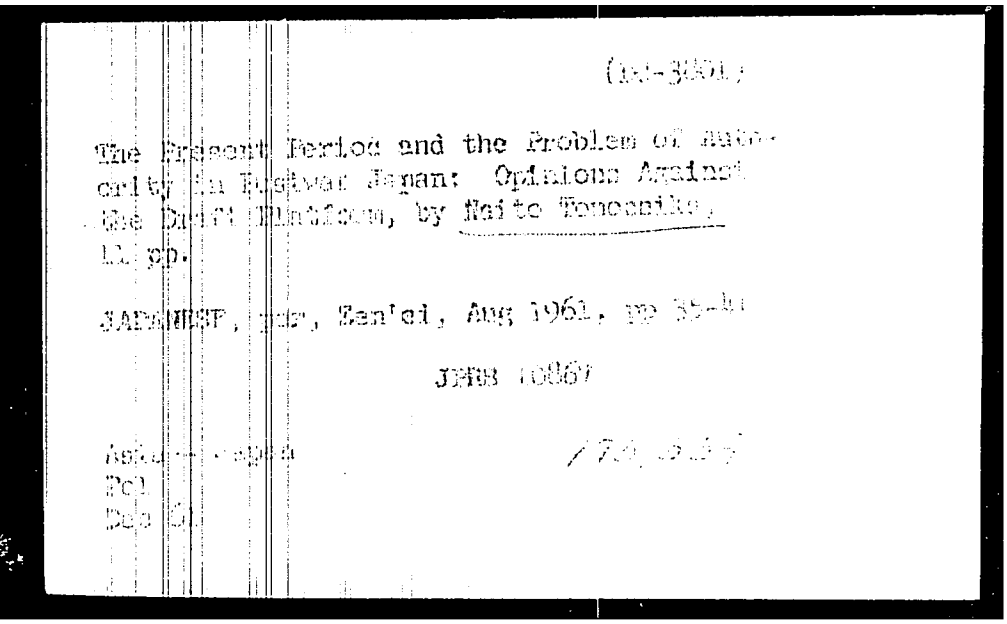

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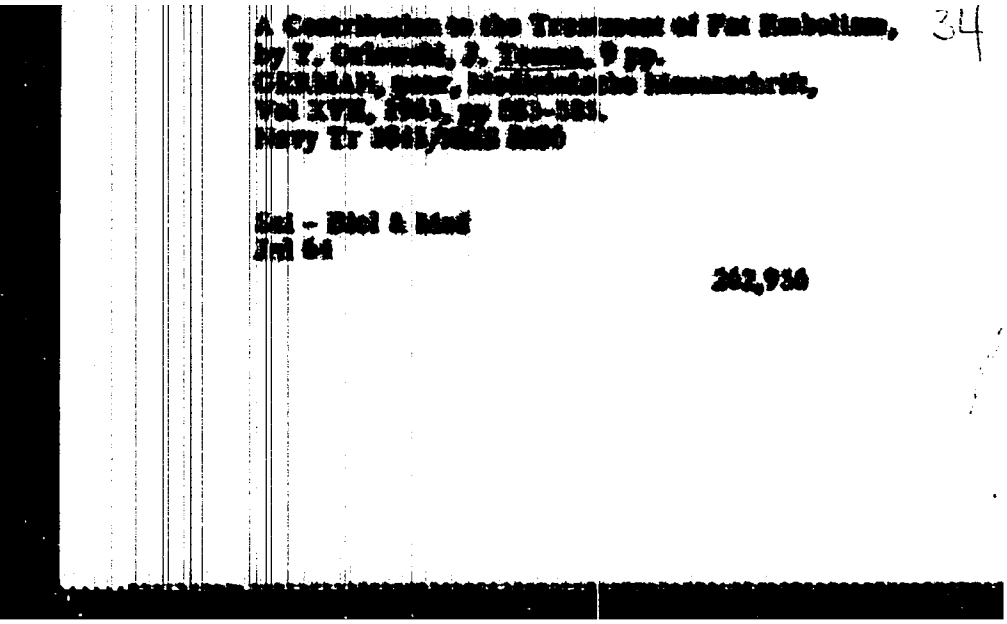

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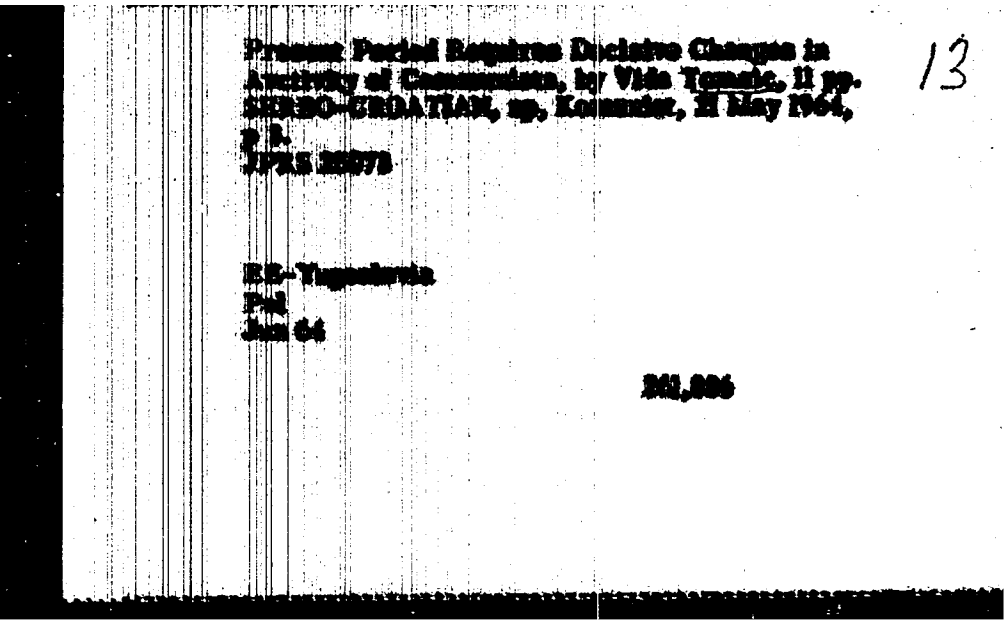

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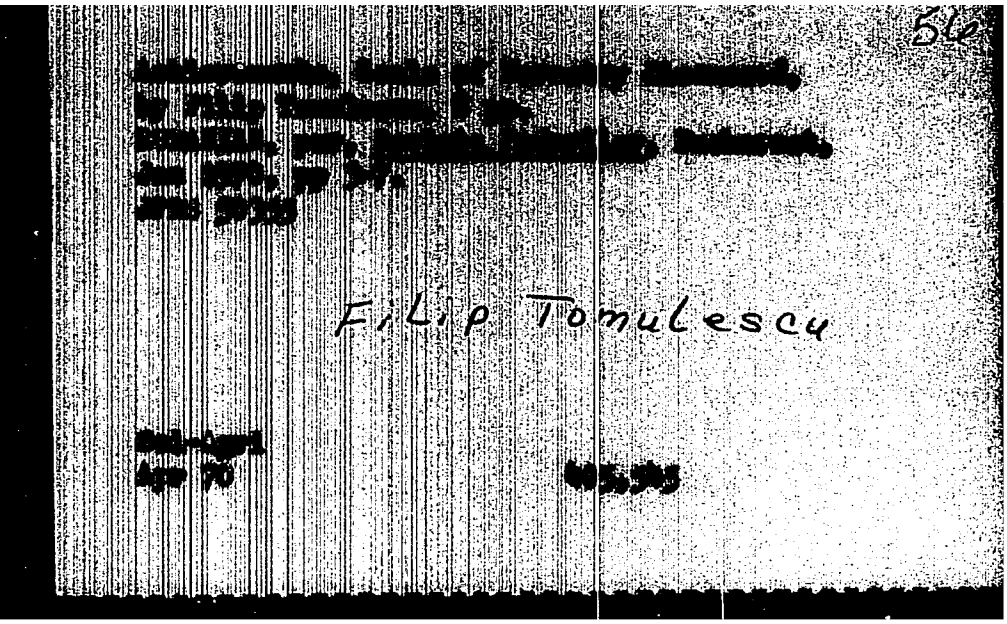

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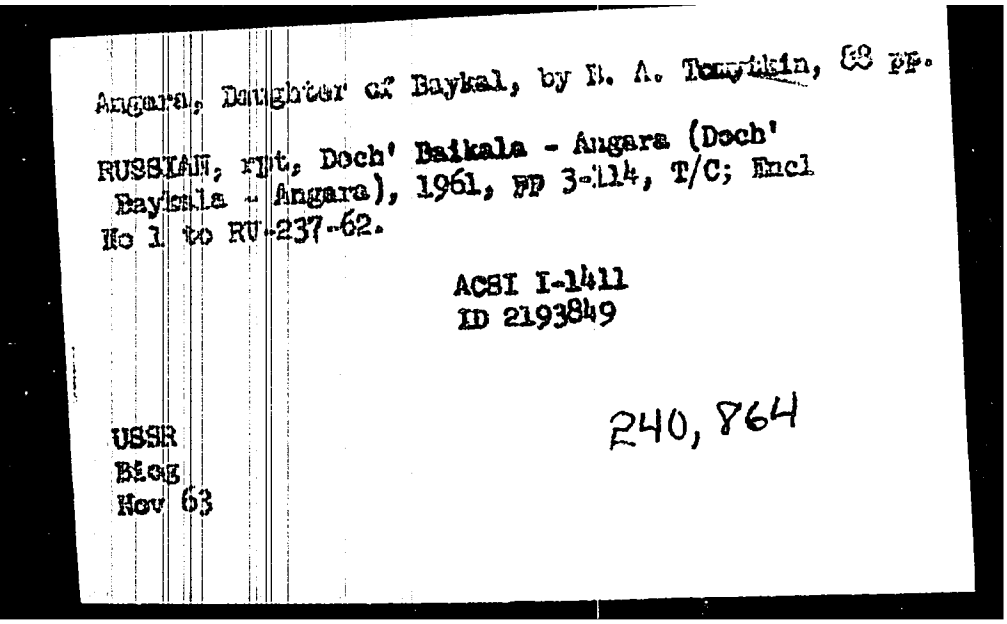

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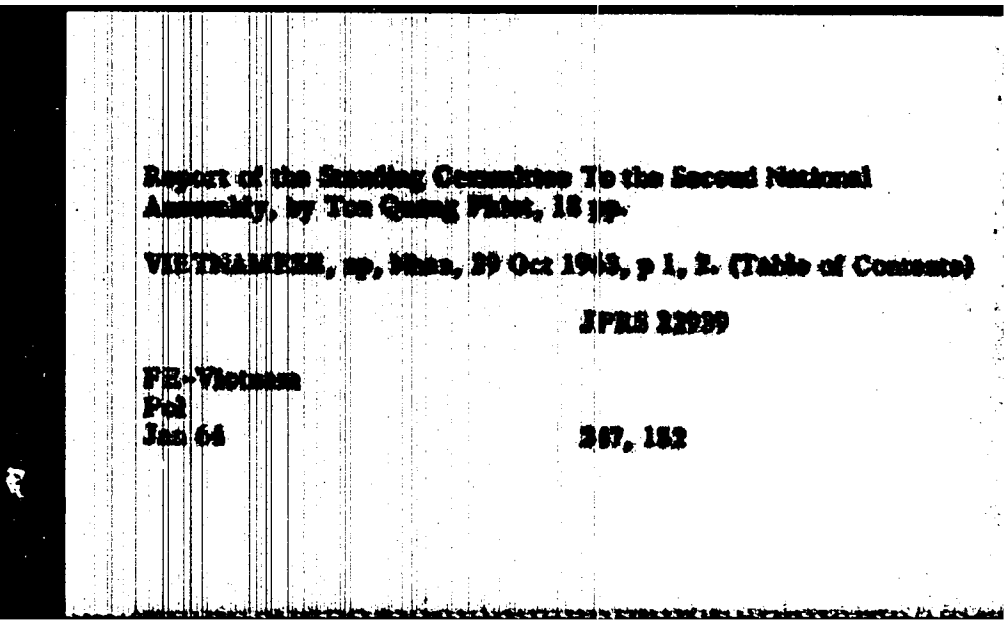

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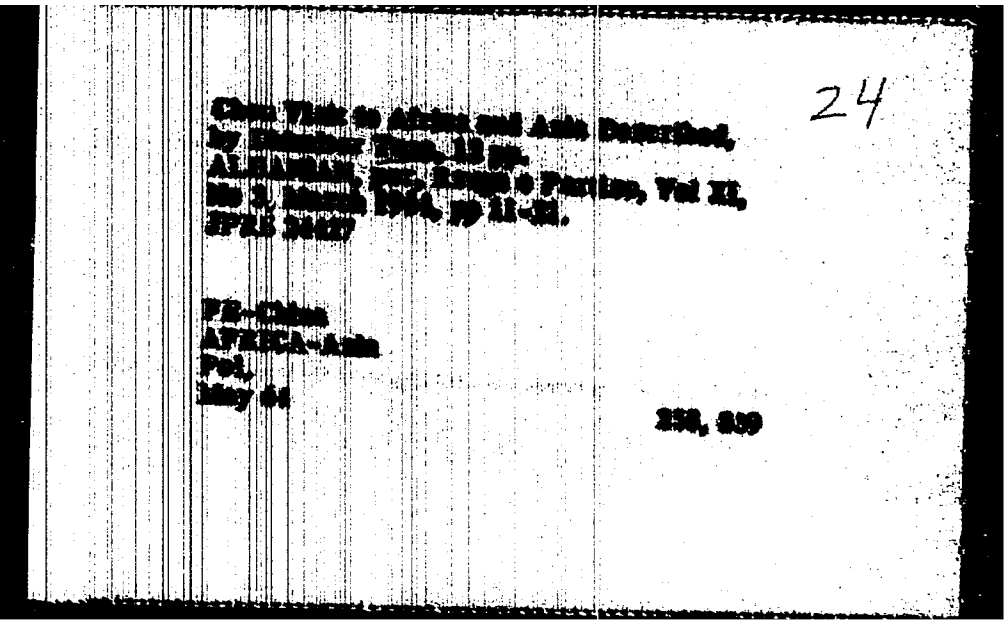

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